

July 2012
U.G. (Diploma) Final Semester Examination

CHEMISTRY
Paper VIII (H) - Physical Chemistry H.S.W.H. 4000

Time 3 hours

(M= Marks)

Note : Attempt all questions. All questions carry equal marks. The blind conditions will be given 60 minutes extra time.

- (a) Discuss in detail the activation complex theory of the reaction rate. In what way is this theory superior to collision rate theory? 6 8½
- (b) Illustrate general features of the fast reactions. Describe flow method to study the fast reactions. 6 8½

OR

- (a) Illustrate the mechanism and kinetics of pyrolysis of acetone-aldehyde. 5
- (b) Illustrate kinetic salt effect with suitable examples. 5
- (c) Describe the mechanism and kinetics of photo-chemical reaction between H_2 and Br_2 . 5
- (d) Describe the Gibbs adsorption isotherm. 6 8½
- (e) What is CMC? Describe various factors affecting the CMC of surfactants. 6 8½

OR

- (a) Write short notes on any two of the following
 i) Surface tension & vapour pressure of droplets (Kelvin equation)
 ii) Adsorption & surface area (BET equation)
 iii) Crystallization of micelles.
 iv) Surfactants & active agent and their classification. 8½ 8½
- (b) Present an illustrative account of liquid crystal polymers and their applications. 8½
- (c) Discuss the kinetics and mechanism of radical chain polymerization. 8½

OR

- (a) Describe the method for the determination of molecular weight of polymers by viscometry. 8
 Equal numbers of molecules with $M_1 = 10,000$ and $M_2 = 100,000$, are mixed. Calculate number average molecular mass and mass average molecular mass. 8½

(b) Explain the Onsager's reciprocity relation. 8

(c) Write short notes on -
 i) Ionic conduction. 3
 ii) The electric kinetic phenomena. 3

OR

- (a) State the principle of minimum entropy production. 5
 Now the non equilibrium stationary states of thermodynamics. 3
 i) Diffusion and Masses. 3 5

- (b) State the theory of colloidography. What is half wave potential? Write its significance. 4
 Answer the following
 i) Colloidal system
 ii) Electro coagulation in context of Polargraphy. 4

OR

- (a) Derive Volmer equation and also explain the Tafel plot. 4
 (b) Discuss Brønsted-Lowry treatment and its extension for conductivity studies of strong electrolytes. 4